Number 8/2

The atmosphere has many minor gases. Some of these are the socalled greenhouse gases (GHGs), which absorb infrared radiation. These heat up the atmosphere, and cause global warming.

## The Role of Greenhouse Gases

## Introduction

The atmosphere is full of minor gases. Some of these are the so-called greenhouse gases (GHGs), which absorb infrared radiation and some heat up the atmosphere. Some gases don't (non-GHGs). Some GHGs absorb infrared radiation more than others.

It is not easy to tell exactly what a GHG is and what is not. However, it is vital to know what are the most important GHGs.

There are two ways of increasing GHGs in the atmosphere:

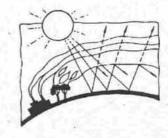
- O increase sources (emitters of GHGs) much of this is man-made.
- O decrease sinks (absorbers of GHGs) much is also man-made.

## Important GHGs

Water vapour is the main GHG for our climate. Levels in the lower atmosphere are determined by the balance between evaporation and rainfall. It is not directly affected by human activity.

Carbon dioxide (CO2) is a very important GHG. Higher concentrations in the atmosphere are a direct result of man-made activities such as emmissions from fossil fuel (coal, gas and oil). In the atmosphere, CO, is very stable and lasts for many decades. About 40% of CO, released into the atmosphere is absorbed by the oceans these are a sink. Some are absorbed in the process of photosynthesis by vegetation on land and by planktons in the sea. However, as plants and plankton decay, CO, is released again into the atmosphere and so only a small amount of CO<sub>2</sub> is permanently removed.

So, CO, in atmosphere is well above its natural levels, so changing normal carbon cycles. This is further affected by forest destruction, and by global warming effects associated with chemical and biological processes in oceans.



Methane (CH<sub>4</sub>) is another important GHG which increases in concentration because of man activities - by converting land into agriculture, by deforestation, and by coal mining and extracting natural gas. Effects of global warming can increase GHGs more when CH, is released from frozen tundra and ice of the arctic regions into the atmosphere.

CH, is short lived compared to CO,, with about 10 years life in the atmosphere. It is destroyed by chemical reactions with other atmospheric substitutes.

Chloro-fluoro carbons (CFC) 11, 12 and 13 are not natural gases. These are man-made, and are emitted from aerosol cans, propellants, refrigerants, foam production and solvents. Their effects last for decades. While levels are rapidly rising it is expected that concentrations should reduce, with the Montreal Protocol requiring Parties to reduce the use of substances that deplete the ozone layer. However, CFC also affect the climate in ways that are still not fully understood.

Ozone (O3) and nitrous oxides (NO) are important GHGs, which are increasing indirectly because of human activity. Ozone in the stratosphere, which protects human beings from the sun's ultraviolet radiation, are being destroyed, so causing an "ozone hole". This is due to reactions, for example, with carbon monoxide (CO) and NO (from car exhaust fumes), CH, and other hydrocarbons. NO is also increasing but for reasons not well understood.

Other GHGs include hydro-chloro-fluoro carbons (HCFCs), hydrocarbons, and halons. Although these gases are less significant, they are becoming important as industrial processes change. HCFC levels are expected to increase rapidly as a result of the Montreal Protocol as it replaces CFC.

CO, increases have dominated the enhanced greenhouse effect to date, but the combined effects of the other gases now contribute to over 40% of the greenhouse ef-



Published by the South Pacific Regional Environment Programme (SPREP), PO Box 240, Apia, Western Samoa, with financial assistance from the United Nations Environment Programme (UNEP). Printed by Commercial Printers, Apia, 1993. Printed on Ecolade recycled paper.